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1 **Production of xylooligosaccharides, bioethanol, and lignin from structural**
2 **components of barley straw pretreated with a steam explosion**

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8 **Abstract**

9 Barley straw (BS) is a potential source to obtain bioethanol and value-added
10 products such as xylooligosaccharides (XOS) and lignin for application in diverse
11 industries. In this study, BS was submitted to steam explosion pretreatment to valorize
12 the main components of this lignocellulose biomass. For hemicellulose fraction
13 valorization, different combinations of endo- β -(1,4)-D-xylanase enzyme with accessory
14 enzymes (α -L-arabinofuranosidase, feruloyl-esterase and acetylxylan-esterase) have
15 been studied to produce XOS with a low degree of polymerization. The application of
16 accessory enzymes combined with endo- β -(1,4)-D-xylanase enzymes turned out to be
17 the most effective strategy for the formation of XOS. The solid fraction obtained after
18 the pretreatment was submitted to presacharification and simultaneous saccharification
19 and fermentation process for bioethanol production. The resulting lignin-rich residue
20 was characterized. In this integrated process, 13.0 g XOS (DP2-DP6), 12.6 g ethanol and
21 16.6 g lignin were obtained from 100 g of BS, achieving the goal of valorizing this
22 agricultural residue.

23 **Keywords:** lignin, bioethanol, bio-refinery, enzymatic hydrolysis,
24 xylooligosaccharides

25 **1. Introduction**

26 The hemicellulose is the second polymer more abundant in the plant cell wall. This
27 term describes diverse heteropolysaccharides, with a variable structure formed of
28 hexoses (D-glucose, D-galactose and D-mannose) and pentose (D-xylose and L-
29 arabinose) (Limayem and Ricke, 2012). The hemicelluloses are classified according to
30 the main residues of sugars present in the backbone of the structural polymer, being
31 xylan is the main hemicellulose in herbaceous plants and hardwoods. In gramineous
32 plants, the xylans are made up of the main backbone of xylose units linked by β -1,4
33 bonds which, shows a great variety of linkages and branching by functional groups such
34 as acetyl, methyl, cinnamic acids (*p*-coumaric or ferulic acid) or glucuronic acids (Gírio
35 et al., 2010). These functional groups usually covalently linked between
36 arabinofuranosyl residues whereas, *p*-coumaric or ferulic acids are joined through an
37 ester linkage.

38 The hemicellulose polymers are interconnected by covalent and hydrogen bonds
39 with other polymers present in lignocellulosic biomass. Via hydrogen bonds, the
40 hemicellulose chains are bound to cellulose and aromatic esters to lignin, joining the
41 bond between cellulose and lignin (Houfani et al., 2020). For accessing carbohydrates,
42 the lignocellulosic biomass must undergo fractionation steps to overcome these forces
43 and thus utilise the main constituents to carry out individualised processing.

44 The pretreatment approach related to hemicellulose is dependent on the final
45 application of hemicellulose. Thus, on this occasion, the steam explosion (SE) method
46 was employed to alter the structure of the lignocellulose biomass. For biomass such as
47 agro-industrial waste, this pretreatment is one of the most excellent for disturbed the
48 structure (Moreno et al., 2019). The feedstock is subjected to high temperature at

49 saturated steam ($> 160^{\circ}\text{C}$) during a period of time before sudden depressurisation. In
50 these conditions; it is produce to hydrolysis of acetyl groups present in the hemicellulose
51 generated acetic acids leading to a greather hydrolyze hemicelluloses (Duque et al.,
52 2016).

53 From hemicellulose fraction in the biorefinery context, high added value products
54 can be obtained to be applied in many manufactures such as pharmaceutical, food or
55 energy. In this context, xylooligosaccharides (XOS) are an interesting compound
56 because they belong to the so-called emerging prebiotics. Prebiotics are defined as “a
57 substrate that is selectively utilised by host microorganisms conferring a health benefit”
58 (Gibson et al., 2017). These compounds have many potential applications in the
59 cosmetic, chemical, or pharmaceutical (Gibson et al., 2017). It is estimated that the
60 prebiotic market will impact 9.5 billion by 2027 (Ahuja and Mamtani, 2021).

61 XOS with a low degree of polymerization presents several biological benefits;
62 they can improve the immunomodulatory functions, antiinflammatory properties, blood
63 sugar reducer and antioxidant activity (Fernández et al., 2016; Gibson et al., 2017;
64 Pinales-Márquez et al., 2021; Slavin, 2013). These biological activities can be affected
65 by molecular weight, distributions, substitutions of XOS or the source and process used
66 for obtaining it (Zhang et al., 2018).

67 In the literature, many lignocellulosic residues have been studied to obtain XOS.
68 Agricultural residues are a promising feedstock for emerging bioeconomic concept
69 (Wietschel et al., 2019). Different lignocellulosic biomasses are explored in the
70 production of XOS, such as chestnut shells, wheat straw or peanut shells (Gullón et al.,
71 2018; Pinales-Márquez et al., 2021; Rico et al., 2018). Among lignocellulosic materials,
72 barley straw (BS) is an interesting residue that can be transformed on by-products

73 interested in many industries. The bioeconomic forecast potential from barley straw has
74 been estimated as 18 million tons in 2030. This estimation is based on considering
75 competing applications of straw (beeding animals, other agricultural uses and mushroom
76 cultivation) (Wietschel et al., 2019).

77 The XOS can be produced mainly by different strategies, chemical and
78 hydrothermal pretreatment, enzymatic hydrolysis (EH) or a combination of pretreatment
79 and EH (Arai et al., 2019; Gullón et al., 2018; Huang et al., 2017; Rico et al., 2018). In
80 EH process are used a xylanolytic enzyme system, which is generally formed of the
81 repertoire of hydrolytic enzymes such as endo- β -(1,4)-D-xylanase, β -xylosidase, α -L-
82 arabinofuranosidase, acetylxylan esterase, and α -D-glucuronidase. The use of
83 xylanolytic enzymatic allows several advantages such as higher yields, mild reaction
84 conditions and no generation of by-products when as compared with the others process
85 (Sadaf and Khare, 2014). To obtain XOS with low degree of polymerization (DP2-DP6),
86 various hydrolytic enzymes are required with different activities, which release the side-
87 chain substitution, achieving major depolymerisation of xylan.

88 The production of ethanol from lignocellulosic biomass based on the use of
89 enzymes followed by fermentation is considered a promising technology in biochemical
90 platforms. Bioethanol production from agriculture residue produces a residue rich in
91 lignin, which can be transformed into surfactants/dispersant additives or aromatic
92 compounds (Santos et al., 2015). Typically other application of this residue is used as an
93 energy resource (Sannigrahi and Ragauskas, 2011).

94 The main goal of this investigation was to study the behavior of several enzymes
95 with accessory activities for increases the production of XOS using SE to solubilize the
96 hemicellulose from barley straw. Also, a brief preliminary study of the integration of

97 bioethanol production from sugars and energy generation from the rich lignin residue of
98 barley straw is presented.

99 **2. Material and methods**

100 **2.1. Raw material and pretreatment**

101 Barley straw (10% moisture) was provided by CEDER (Centro de Desarrollo de
102 Energías Renovables, Spain). The biomass composition was determined according to the
103 NREL procedures (Sluiter et al., 2010). BS (6-10 mm) was pretreated by the steam
104 explosion in a 10 L reactor at 180 °C for 30 min in a small prototype plant (CIEMAT, Spain).
105 The reactor was preheated at the set pretreatment temperature with saturated steam.
106 After biomass addition into the reactor, it took less than 60 seconds to reach working
107 temperature. The pressure necessary to reach the temperature was 10 bar. These
108 conditions were chosen as such compromise conditions to recover more amounts of
109 cellulose and hemicellulose as possible. After the steam explosion pretreatment, the
110 material (slurry) was recovered, cooled and filtered to separate insoluble solids and the
111 liquid fraction. The solid fraction were thoroughly washed obtaining the water insoluble
112 solids fraction (WIS). The liquid fraction can contain degradation compounds from
113 sugar and derivates lignin, were removed using a cleaning step with ion exchange resin
114 (Microionex MB200 resin (Rohm-Hass Copenhagen, Denmark), following the method
115 Negro et al. (2014) giving rise to a sample called liquid fraction of barley straw
116 pretreated (*Lfb_{sp}*). The liquid from the steam explosion pretreatment, free of degradation
117 compounds, will be called the degradation compounds will be referred to as the liquid
118 fraction of pretreated barley straw (*Lfpbs*), and will be the starting substrate to obtain the
119 low polymerization degree xylooligosaccharides.

120 **2.2. Enzymatic hydrolysis of the liquid fraction of barley straw pretreated**

121 A series of enzymes with different activities were chosen to obtain XOS with
122 (DP2-DP6) from the *Lfbsp*. Two Commercial endo- β -(1,4)-D-xylanase M1 (E-XYTRI
123 M1 GH11) from Megazyme International (Bray, Ireland) and *NS50030* from
124 Novozymes AS, (Bagsværd, Denmark), were used. In addition, accessory enzymes that
125 are able to hydrolysate linkages among xylan chain and substituent residues such as α -L-
126 arabinofuranosidase (E-AFASE, GH51), feruloyl esterase (E-FAEZCT, CE1), and
127 acetylxylan esterase (E-AXEAO-IKU, CE6) from Megazyme International (Bray,
128 Ireland) were tested. Enzyme loadings for the several tests were following: *Endo- β -*
129 *(1,4)-D-xylanase* M1 5 U/mL; *Endo- β -(1,4)-D-xylanase NS50030* 7.2 U/mL ; α -L-
130 arabinofuranosidase 6.3 U/mL; feruloyl esterase 0.05 U/mL, and acetylxylan esterase 5
131 U/mL. The reactions were stopped by boiling at 100°C for 5 minutes before being
132 analysed. The reaction conditions were the following pH 4.8 and 50°C (0.05 M sodium
133 citrate buffer) in an incubator shaker at 150 rpm (Infors HT Minitron Bottmingen,
134 Switzerland).

135 **2.3. Presaccharification simultaneous saccharification and fermentation process** 136 **(PSSF) of WIS**

137 A PSSF process was carried out as previously described Álvarez et al. (2018).
138 Briefly, the pre-hydrolysis stage (8 h at 50°C) was performed using the WIS obtained
139 after pretreatment at a solids concentration of 20% (w/v) and an enzyme load (Cellic
140 CTec2 Sigma Aldrich USA) of 30 FPU/g glucan. The fermentation stage was carried out
141 at 35°C using *Saccharomyces cerevisiae* Ethanol Red (inoculum size 0.25 g/L). After 48
142 h of process, the resulting residue was washed with water and dried at 60°C, determining
143 the content of carbohydrates and lignin (Sannigrahi and Ragauskas, 2011) .

144 **2.4. Analytical methods**

145 *2.4.1. Characterization of raw material, WIS and lignin-rich residue*

146 The chemical composition of raw material, WIS and the lignin-rich residue
147 was determined using the method described by Sluiter et al. (2010).

148 Additionally, in the lignin-rich residue, the ultimate analysis was measured by
149 using LECO TRUSPEC analyser, following the procedure based on ISO 1648. Calorific
150 value was determined by utilizing a LECO AC500 bomb calorimeter.

151 *2.4.2. Sugar (oligomers and monomers), furans, phenolic compounds and* 152 *organic acids*

153 As the sugars (monomeric and oligomeric form), such as sugar degradation
154 compounds included furfural and 5- hydroxymethylfurfural (HMF), phenolic
155 compounds (ferulic and *p*- coumaric acids) from lignin released during pretreatment and
156 acetic acid were analysed and quantified by high-performance liquid chromatography
157 HPLC following the methodology described in Álvarez et al. (2017).

158 *2.4.3. Xylooligosaccharides analysis*

159 XOS X₂, X₃, X₄, X₅ and X₆ (xylobiose, xylotriose, xylotetraose, xylopentose and
160 xylohexose, respectively) generated in the process were quantified by High performance
161 anion-exchange chromatography ICS2500 Dionex System (Dionex Corporation,
162 Sunnyvale, CA) with a pulsed amperometric detector (ED50 electrochemical detector)
163 with a gold electrode using Dionex CarboPac PA100 analytical column and PA100
164 guard-column. The sample were eluated with a linear gradient program from 100% A
165 (NaOH 100 mM) to 30% B (NaOAc 500 mM/NaOH 100 mM) and at flow rate of 0.6
166 mL/min.

167 *2.4.4. Fourier Transformed Infrared Radiation analysis*

168 FTIR spectra Fourier Transformed Infrared Radiation (FTIR) spectra were
169 measured in a Nicolet (Thermo Fisher Scientific) Nexus spectrometer equipped with a
170 Smart Golden Gate ATR (Thermo Electron Scientific Instruments LLC, Madison, WI
171 USA) accessory attenuated total reflectance device. Spectra were collected in the 400-
172 600 cm^{-1} range with a 0.25 cm^{-1} resolution and an average of 64 scans. This technique
173 was used for the characterization of lignin-rich residue.

174 **3. Results and discussion**

175 **3.1. Characterisation of raw material and fractions obtained after** 176 **pretreatment**

177 Cellulose, hemicellulose, and acid insoluble lignin content on barley straw
178 represent 32.9%, 27.2%, and 16.8% dry matter content, respectively (Fig. 1). The
179 majority component of hemicellulose was xylan (22.1%), and minor proportions were
180 determined as arabinans (3.6%), galactans (1.3%), and mannans (0.3%). Acetyl groups
181 account for 1.7%. Although there are slight variations, due to various factors such as the
182 variety of barley, soil composition, climate or type and amount of fertilisers used, and
183 also the method of chemical composition characterisation, the barley composition is in
184 the range of those determined by other authors, 32-40% in cellulose, 21-27% in
185 hemicellulose and 15-22% in lignin (Duque et al., 2014b, 2014a; Lara-Serrano et al.,
186 2018; Sáez et al., 2013). Above 60% of compounds present in the lignocellulosic
187 material of BS are carbohydrates, an excellent material for sugars production and
188 conversion to ethanol, and high added-value products such as producing XOS (DP2-
189 DP6). It is an appropriate substrate to be used in biorefineries.

190 The compositions of solid obtained after pretreated are depicted in figure 1. This
191 fraction contained a higher proportion of cellulose (55.8%), and lignin (30.7%) than raw
192 material (Fig. 1). Pretreatment increased the cellulose proportion of WIS due to the
193 extensive solubilisation of the hemicellulose fraction. Hemicellulose content was 10.3%.

194 ***Fig 1.***

195 In liquid fraction, sugars were determined as monomeric and oligomeric form,
196 being the xylose (monomeric form, 3.1 g/L as oligomeric form 25.3 g/L) the main sugar,
197 followed of glucose (0.9 and 5.2 g/L, and minor amounts galactose 1.6 and 0.6 g/L,
198 arabinose 1.2 and 2.1 g/L and mannose 0.4 and 0.07 g/L such monomeric and
199 oligomeric form respectively.

200 During pretreatment, in addition to solubilising the hemicellulosic-derived
201 sugars, acetyl groups were also released from hemicellulose in the form of acetic acid
202 (1.7 g/L). Also, a small concentration of compounds derived from lignin such as vanillin
203 (4-hydroxy-3-methoxybenzaldehyde) (40.5 mg/L) from the release of guayacyl propane
204 units, and ferulic acid (34.5 mg/L), and *p*-coumaric acid (30 mg/L) compounds that are
205 part of the group of hydroxycinnamic acids were detected. Finally, degradation products
206 such as furfural and HMF (0.9 and 0.1 g/L) had also been identified, respectively.

207 Liquid fraction (hydrolysate) contained soluble hemicellulose to obtain high-
208 added value compounds, such as prebiotic xylooligosaccharides. Most of the
209 degradation compounds (furfural, HMF), acetic and formic acids and phenolic
210 compounds generated during pretreatment were removed using an ion exchange resin as
211 cleaning step. The sugars value was not affected by the cleaning step. However, the
212 concentration of furans compounds were reduced up to 0.22 and 0.02 g/L furfural and HMF

213 respectively, while those carboxylic acids were eliminated. This step is necessary because
214 these degradation compounds may hinder enzymatic hydrolysis. *Lfbsp* presented an
215 oligosaccharides concentration of 31.7 g/L, of which 25.3 g/L are xylooligosaccharides.
216 These values refer to raw material are equivalent to 17.5 g oligosaccharides/100 g barley
217 straw and 13.9 g XOS/100 g BS.

218 XOS (DP2-DP6) profile present in the *Lfbsp* is X₂ (2.8 g/L), X₃ (2.5 g/L), X₄ (2.8
219 g/L), X₅ (2.7 g/L), and X₆ (2.4 g/L) and account for 7.9 g/100 g of raw material. Several
220 enzymes with different activities (endo- β -(1,4)-D-xylanase, α -L-arabinofuranosidase,
221 acetylxylan esterase and feruloyl esterase) has been studied in order to achieve the
222 depolymerisation of xylan chains present in the liquid fraction. This set of enzymes
223 forms an enzymatic cocktail. The advantage of using an enzymatic cocktail consortium
224 is that it acts on the terminal and/or internal glucosidic linkages; each enzyme can act on
225 different bonds present in the polymers.

226 **3.2. Enzymatic hydrolysis of barley straw pretreated for production of XOS**

227 In previous work carried out, it has chosen of the pure endo- β -(1,4)-D-xylanase
228 M1 (GH11) from *Trichoderma viride* as the endoxylanase that more amount of XOS
229 produce from barley straw pretreated in front endo- β -(1,4)-D-xylanase from *Thermotoga*
230 *maritima* GH10 (Álvarez et al., 2018).

231 In order to progress the research, the next step was to add accessory enzymes for
232 hydrolysis of hemicellulose. In this section, the results achieved in the experiments carry
233 out with the combination of enzymes were showed. Four different enzymes formulations
234 were evaluated for increased XOS from BS pretreated.

235 **3.2.1. Enzymatic hydrolysis employing endo- β -(1,4)-D-xylanase (M1), α -L-** 236 **arabinofuranosidase, and feruloyl esterase**

237 The first cocktail of the enzyme used was called *Maf*. This cocktail is composed of
238 endo- β -(1,4)-D-xylanase (M1), α -L-arabinofuranosidase, and feruloyl esterase. The
239 goals of this enzymatic hydrolysis are *i*) help hydrolysis of xylanase, a obtain XOS of
240 low degree of polymerisation and *ii*) reduce the substitutes presents in XOS since as
241 more lineal is XOS easier fermentable will be for bifidobacteria (Vazquez-Olivo et al.,
242 2019).

243 The efficiency of enzymatic hydrolysis can be reduced through arabinoxylan with
244 ferulic acid cross-linking (de Oliveira et al., 2015). In this context, the synergy action of
245 feruloyl esterase has been demonstrated with endo- β -(1,4)-D-xylanase) and α -L-
246 arabinofuranosidase, among others in the degradation of plant cell walls. This fact
247 supposes a reduction of the enzyme dosage, consequently impacting the prices of the
248 final products. The concentration of cinnamic acids (ferulic acid and *p*-coumaric) was
249 monitoring during EH trial. The maximum value of the concentration of ferulic and *p*-
250 coumaric acid (96 and 29 mg/L respectively) was achieved in a short time of hydrolysis
251 (1 h of reaction). From this time on, the concentrations remained constant. The
252 production of ferulic and *p*-coumaric acids is because the enzyme feruloyl esterase can
253 hydrolysing the ester linkage (Wong et al., 2013).

254 In addition, the production of low DP XOS was monitored over reaction time
255 (Fig. 2a). The XOS majority was xylobiose, whose maximum value is 13.6 g/L at 5 h.
256 This concentration supposes 48.8% of potential xylose present as oligosaccharide in the
257 liquid fraction of barley straw pretreated. The maximum amounts of X₂, X₃, X₄, X₅ and
258 X₆ were 12.5; 3.5; 2.6; 1.4; and 0.7 g/L, respectively. These values were achieved at 4 h
259 of reaction, and these data have shown the ability of the enzyme cocktail to produce
260 mainly short-chain XOS (xylobiose and xylotriose). A similar trend was observed by

261 Avila et al. (2020b) in the hydrolysis of extracted xylan from sugarcane straw and
262 coffee husk by a mixture of endo- β -(1,4)-D-xylanase, α -L-arabinofuranosidase, and
263 feruloyl esterase. XOS with a low degree of polymerisation are easily metabolised by
264 probiotic bacteria, being these types of XOS preferred for food applications industries
265 (Reddy and Krishnan, 2016).

266 **Fig. 2.**

267 Concerning arabinose production, the release of arabinose (0.4 g/L) was due to the
268 action of α -L-arabinofuranosidase together with feruloyl esterase because some
269 arabinofuranosyl residues are esterified at position 5 with ferulic acid (Biely et al.,
270 2016). Despite this, 65.8% of arabinose continues to link to the soluble hemicellulose.
271 This fact, of the incomplete release of arabinose, could present other substituents such as
272 acetyl groups present in the hemicellulose. These groups can be hampering the action of
273 enzymes.

274 In this assay, 12.3 g de XOS (DP2-DP6) per 100 g of BS was obtained, meaning an
275 increase of 56% concerning *Lfb*sp.

276 **3.2.2. Enzymatic hydrolysis employing endo- β -(1,4)-D-xylanase (M1), α -L-** 277 **arabinofuranosidase, acetylxylan esterase and feruloyl esterase**

278 The presence of acetyl groups in the oligosaccharides present in the liquid fraction
279 of BS pretreated was confirmed through techniques FTIR and Matrix-Assisted Laser
280 Desorption/Ionization-Time of Flight-Mass Spectrum in previous work (Álvarez et al.,
281 2020). Thus, acetylxylan esterase enzyme was added to a before cocktail enzymatic,
282 refer to *Complex* cocktail to increase the production of XOS (DP2-DP6). Acetylxylan
283 esterases present a synergistic action with GH10 and GH11 xylanases (Zheng et al.,

284 2013). The acetyl groups present in the hemicellulose hampered the action of endo- β -
285 (1,4)-D-xylanase enzymes limiting the degree of hydrolysis.

286 The SE pretreatment was not able to fully be released acetyl groups present in the
287 hemicellulose. In the depolymerisation of solubilised xylan, the acetylxylan esterases
288 enzymes liberate acetic acid, esterifying D-xylopyranosyl residue. The yield of acetic
289 acid production was relatively low (33.7%), even though the incorporation of
290 acetylxylan esterase. The incomplete deacetylation that occurred in this assay can
291 explain several hypotheses. The experiment was carried out in a compromise condition
292 pH (4.8) for an enzymatic cocktail. This pH is very different from the optimum pH for
293 acetyl esterase (pH 7). This fact can be reduced the mode of the act of acetyl esterase.
294 Another theory for just the low deacetylation can be due to steric hindrance. The
295 literature describes the existence of oligosaccharides with acetylated resistant groups, in
296 particular acetyl groups located on non-reducing-end xylopyranosyl residues that may
297 become non-hydrolysable by spontaneous migration from position 2 and 3 towards
298 position 4 so that total deacetylation does not occur (Biely et al., 2013).

299 In this test, the maximum release of cinnamic acids (ferulic and *p*-coumaric acids)
300 occurred at short reaction times (1 h). With the *Complex* enzymatic combination, the
301 release has been somewhat lower, obtaining maximum values of 88 and 22 mg/L of
302 ferulic and *p*-coumaric acids, respectively, and this fact could be attributed to the
303 impediment that the presence of the four enzymes could produce. Wu et al. (2017)
304 observed a limited action of feruloyl esterase when increasing the concentration of endo-
305 β -(1,4)-D-xylanase since that the xylan substituents prevent hydrolysis.

306 Figure 2b illustrates the production profile of different XOS (X₂, X₃, X₄, X₅, and
307 X₆) obtained using the *Complex* as an enzymatic cocktail. During the enzymatic

308 hydrolysis, a significant concentration of XOS (DP2-DP6) was produced at brief
309 reaction times (45 minutes). After this time, xylose and xylobiose concentrations were
310 increased to a value of 4.5 and 13.7 g/L, respectively, at 5 h. This increase is principally
311 due to a decline in XOS with DP3, DP4, DP5, and DP6 present in the medium. This fact
312 was produced because eliminating the acetyl groups increases the areas in which the
313 enzyme can act by releasing XOS of a low degree of polymerisation (Zhang et al.,
314 2011). The concentration of X₃ was increased to 6.6 g/L at 45 minutes of EH, from
315 which it decreases up to a value of 3.8 g/L. In the case of xylotetraose and xylopentose,
316 the behaviour was similar. The maximum concentration was achieved a brief time (45
317 minutes), for later decreasing in favour of XOS with a minor degree of polymerisation.
318 In the case of xylohexose was hydrolysed from the beginning of the reaction to a value
319 of 0.1 g/L.

320 The side chains may influence the rate of hydrolysis of xylan (Brienzo et al.,
321 2016); in this case, the hydrolysis is faster due to the presence of hemicellulose,
322 accessory enzymes acting on the backbone branches.

323 Given these conditions, 13.0 g of XOS (expressed in xylose form) with DP2-DP6
324 per 100 g of BS were produced. In this case, an incremented of 64% concerning *Lfbsp*
325 was achieved.

326 **3.2.3. Enzymatic hydrolysis using a commercial endo- β -(1,4)-D-xylanase**

327 Since supplementation with accessory enzymes to hydrolysis with pure
328 commercial endo- β -(1,4)-D-xylanase has recovered high depolymerisation of the pre-
329 hydrolysate, favouring of increased XOS formation, EH experiments have been carried
330 out with an endo- β -(1,4)-D-xylanases commercial enzyme (*NS50030*). This enzyme

331 presents various activities such as β -xylosidase, α -L-arabinofuranosidase, endo-glucanase,
332 among others.

333 The action of endo- β -(1,4)-D-xylanase *NS50030* on the *Lfbsp* produced the
334 maximum amount of DP2 to DP6 oligosaccharides at 5 h of hydrolysis, equivalent to a
335 production yield of 88.0%. This yield is like that obtained in the *Maf* assay (88.4%),
336 though in the shortest time. This result may be because the enzyme endo- β -(1,4)-D-
337 xylanase *NS50030* also exhibits α -L-arabinofuranosidase activity (Nieto, 2017).

338 As in using pure enzymes, xylobiose and xylotriose were the majority XOS
339 generated after hydrolysis. The concentration of X₂ increased throughout enzymatic
340 hydrolysis until it reached a maximum concentration of 13.8 g/L, while the
341 concentrations of X₃, X₄, X₅, and X₆ decreased (Fig. 3a). It is also observed the increase
342 in X₃ production during the first two hours of EH. After this time, X₃ begins to decrease
343 in concentration, increasing the xylobiose concentration. The same commercial endo- β -
344 (1,4)-D-xylanase was used combined with arabinofuranosidase to obtain XOS from a
345 mixture of sugarcane bagasse and straw for Ávila et al., 2020a. Several pretreatments
346 (ionic liquid and dilute sulfuric acid) were studied in this work, being xylobiose the main
347 product in both pretreatments.

348 **Fig. 3**

349 The maximum amount of XOS achieved was 12.3 g (expressed as xylose)/100 g of
350 raw material, which was achieved at 5 h of hydrolysis. This same enzyme dose (7.2
351 U/mL) was used on pretreated wheat straw (Álvarez et al., 2017), reaching a value of 8.9
352 g XOS/100 g wheat straw. The lower yield obtained in wheat straw is mainly associated
353 with partial degradation of the hemicellulose fraction due to the high temperatures used
354 in the pretreatment (200 °C).

355 About the release of xylose, glucose and arabinose, the amounts obtained
356 represent 5.6%, 25.4% and 76.9%, respectively, of those potentially present, due to the
357 action of the accessory activities *NS50030* enzymatic preparation.

358 **3.2.4. Enzymatic hydrolysis using a commercial endo- β -(1,4)-D-xylanase, α -** 359 **L-arabinofuranosidase, acetylxylan esterase and feruloyl esterase**

360 Like the previous case, the aimed to study the behaviour of complementary
361 enzymes (α -L-arabinofuranosidase, acetylxylan esterase and feruloyl esterase) with
362 commercial endo- β -(1,4)-D-xylanase *NS50030* under the same conditions under which
363 the previous assay was carried out. This cocktail enzymatic is called *NSAFA*.

364 The action of the enzyme acetylxylan esterase released a concentration of acetyl
365 groups from the acetylated oligosaccharides, measured as acetic acid of 0.6 g/L, which is
366 equivalent to 41.2% of the potential acetic acid (referred to the value measured after
367 mild acid hydrolysis with sulfuric acid). This value is lower than expected; this could be
368 because the hydrolysis conditions were not optimal for acetylxylan esterase in terms of
369 pH conditions. The hydrolysis tests under conditions close to the optimum pH for the
370 acetylxylan esterase enzyme resulted in a yield of more than 85% release of the acetyl
371 groups.

372 The effect of incorporating feruloyl esterase enzyme in the hydrolysis medium
373 with *NS50030* was a progressive release of *p*-coumaric and ferulic acids throughout the
374 hydrolysis time. The highest concentration released was 100 mg/L of ferulic acids and
375 49 mg/L of *p*-coumaric. The release of these lignin derivatives did not result in an
376 appreciable increase in arabinose to which (1 \rightarrow 5) bonds link them.

377 Xylobiose was the majority XOS produced throughout commercial EH endo- β -
378 (1,4)-D-xylanase along with accessory enzymes (Fig. 2b). X₂ was increasing to a
379 concentration of 15.7 g/L at 6 h. Xylotriose concentration increased up to 1 h of
380 hydrolysis with a value of 7.3 g/L from which it decreased to values of 2.3 g/L at 6 h.
381 The remaining XOS of X₄, X₅ and X₆ were hydrolysed and transformed to xylotriose.
382 Despite the increase in xylobiose at 6 h, the highest concentration of XOS (DP2-DP6)
383 was produced at 3 h of hydrolysis, obtaining 12.1 g of XOS (DP 2-DP6) expressed as
384 xylose/ 100 g of barley straw. This value is similar to that obtained in the assay of
385 *NS50030* individually, which leads to the conclusion that, under the conditions studied,
386 the complementary enzymes α -L-arabinofuranosidase, acetylxylan esterase and feruloyl
387 esterase do not help the β -(1,4)-D-xylanase *NS50030* enzyme in the release of XOS with
388 (DP2-DP6).

389 Figure 4 shows the total concentration of XOS (DP2-DP6). The highest
390 concentrations of XOS (DP2-DP6) were released by *Complex* enzymatic hydrolysis.
391 This cocktail enzymatic produced 8.6 g/L of XOS in a short time (less than one hour).
392 At the same time, the other assays produced almost 1.5 g/L lower than the *Complex*
393 cocktail.

394 **Fig. 4.**

395 The literature showed a great deal of paper that used endoxylanase to produce
396 XOS from agriculture residues. Goldbeck et al. (2016) studied the effect of a mixture of
397 recombinant endoxylanase and a feruloyl esterase using sugar cane bagasse as raw
398 material. These authors achieved 356 mg of XOS/g xylan. This value is lower in
399 comparison with the results obtained in the assays carried out in this work (557; 588;
400 and 547 mg of XOS/g xylan in *Maf*; *Complex* and *NSAFA*, respectively). However, Liu

401 et al., (2018) evaluated XOS production in corn cobs using *Paenibacillus barengoltzii*
402 xylanase. The yield obtained was 750 mg XOS /g xylan. Although this value is higher
403 than obtained in this work, our enzymatic hydrolysis achieved XOS with a degree of
404 polymerisation between DP2-DP6 since Liu obtained DP2-DP4. The use of endo- β -
405 (1,4)-D-xylanase commercial and α -L-arabinofuranosidase in a sugarcane bagasse
406 produced 338 mg of XOS/ g xylan extracted (Ávila et al., 2020b). In this case, the
407 amount is smaller than achieved in our assays of EH.

408 Several feedstocks such as corncob, rice straw, and almond shells also were used
409 for producing XOS. In these cases, in a range between 180-110 mg of XOS from g of
410 biomass were obtained because of enzymatic hydrolysis using several enzymes and
411 pretreatment (Han et al., 2020; Le and Yang, 2019; Singh et al., 2019). These studies
412 suggest that accessory enzymes act synergistically with endo- β -(1,4)-D-xylanase
413 enzyme during the EH of barley straw pretreated to produce the relevant products such
414 as XOS. The difference of results between *Complex* assays and *NSAFA* is due to the
415 mechanism of heterosynergism. In this case, the heterosynergism occurred for the initial
416 activity of the main depolymerising endo- β -(1,4)-D-xylanase, which produce the
417 substrate for the auxiliary enzyme.

418 **3.3. PSSF process**

419 The PSSF process was carried out with the WIS fraction at optimum conditions
420 based on previous work (Álvarez et al., 2018). An ethanol concentration close to 50 g/L
421 was obtained after 48 h, the ethanol yield was 22.3 g ethanol/g solid pretreated. Lara-
422 Serrano et al. (2018) obtained similar yields in BS pretreated with ionic liquids (22.9
423 g/100 g pretreated straw). However, in this case, and given the type of pretreatment,

424 there has been no separation of the hemicellulose fraction to obtain
425 xylooligosaccharides.

426 *S. cerevisiae* Ethanol Red, the yeast used, exclusively ferments the hexoses to
427 ethanol so that all the xylose produced in the enzymatic hydrolysis stage can not be
428 fermented to ethanol, leaving at the end of the process a final concentration in the
429 fermentation medium around 4-6 g/L of xylose. This xylose could be fermented together
430 with glucose if a co-fermentation strategy was used using microorganisms capable of
431 fermenting both sugars, slightly increasing the final ethanol yield of the process. The
432 overall ethanol production yield in this work was 12.6 g ethanol/100 g of BS, higher
433 than that obtained by Duque et al. (2014), which reported overall yield values of 11.0 g
434 ethanol/100 g of raw material using barley straw and a combined alkali and extrusion
435 pretreatment. More recently, higher production (15.8 g ethanol/100 g of raw material)
436 was reported from barley straw pretreated by combined alkaline and enzyme-catalysed
437 extrusion in a simultaneous saccharification and co-fermentation process genetic
438 modified *S. cerevisiae* strain (Duque et al., 2020). In this case, the hemicellulose fraction
439 was also used for ethanol production.

440 Other authors also used a wheat straw has been submitted to produce XOS and
441 bioethanol. For example, Huang et al. (2017) achieved similar cellulose conversion to
442 ethanol at the present work (0.41 g ethanol/ g cellulose vs 0.42 g ethanol/ g cellulose).
443 However, the fermentation time was less than 48 h in our study, obtaining higher
444 volumetric productivities. Huang et al. (2017) achieved 229 mg of XOS per gram of
445 xylan (mainly xylobiose and xylotriose) using endoxylanase. This value is less than
446 obtained in this work.

447 **3.4. Characterisation of lignin-rich residue**

448 The chemical composition of the lignin-rich residue obtained after the PSSF
449 process is depicted in figure 1. In this residue, the main component was lignin showing
450 content of 74.5%, as expected. In the literature are described similar values for red
451 maple (76%) or olive tree pruning (77%), or such lower as is switchgrass (59%) or
452 wheat straw (69%) (Sannigrahi and Ragauskas, 2011; Santos et al., 2015). However, in
453 the lignin-rich residue, a certain amount of cellulose and hemicellulose were
454 quantified. The composition is carbon 54.1%, hydrogen 6.1% and nitrogen 2.4%.
455 Given the high proportion of lignin in the residue and also its high heating value (23.4
456 MJ/kg, HHV dry basis) this residue proves a suitable green energy source or a
457 precursor for lignin-based fuels and biomaterials. This lignin can be used in many
458 applications such as energetic sources to improve the economy of the process.

459 The residue solid of lignin after simultaneous saccharification fermentation
460 contains a high value of lignin. This residue has been characterised using FTIR
461 spectroscopy (see the supplementary material).

462 A strong, wide band was shown in the spectrum, between 3500 and 3100 cm^{-1} .
463 These bands are assigned to OH stretching vibrations by the presence of alcoholic and
464 phenolic hydroxyl groups involved in hydrogen bonds. The region between 2920 cm^{-1}
465 and 2850 cm^{-1} is assigned to C H stretching vibrations in methyl and methylene groups.
466 The intensity of bands 1510 cm^{-1} has contributed to aliphatic contribution, including
467 carbohydrates and aliphatics side chains in lignin, concerning the aromatic part
468 (Rossberg et al., 2015). In this spectrum, typical band attributed to polysaccharides such
469 as at 1740 cm^{-1} the C=O stretching vibrations in the ester or carboxyl acid groups or 890
470 cm^{-1} β -Glycosidic linkages in pyranose were weak observed. The shoulders at 1719
471 cm^{-1} originate from unconjugated and carbonyl stretches.

472 Aromatic skeletal vibrations give three strong peaks at 1595, 1509 and 1419 cm^{-1}
473 (Sun et al., 2011). These bands are the most characteristic vibrations of lignins (Nunes
474 and Pardini, 2019). Other bands assigned were: 1458 cm^{-1} (asymmetric C–H
475 deformations), 1356 cm^{-1} (symmetric C–H bending), 1319 cm^{-1} (syringyl ring breathing
476 with C–O stretching), 1265 cm^{-1} shoulder (guaiacyl ring breathing with C O stretching),
477 1090 cm^{-1} (C–O deformation, secondary alcohol and aliphatic ethers), 1030 cm^{-1}
478 (aromatic C–H in-plane deformation plus C–O in primary alcohols, guaiacyl type), 1321
479 cm^{-1} (symmetric C–H bending), 1265 cm^{-1} shoulder (guaiacyl ring breathing with C O
480 stretching), 1086 cm^{-1} (C–O deformation, secondary alcohol and aliphatic ethers).

481 The spectra have a stronger wide band at 1030 cm^{-1} than that band at 1136 cm^{-1}
482 indicate the content of the guaiacyl (G) unit is higher than that of the syringyl (S) unit
483 with high condensation (Sun et al., 2011). Also, the band at 901 cm^{-1} indicated aromatic
484 C–H out-of-plane bending in G units, whereas the band at 823 cm^{-1} represented C–H
485 out-of-plane bending in positions 2 and 6 of S units and all positions of *p*-hydroxyphenyl
486 units

487 These results and the high lignin content in this residue offer the possibility of
488 valorisation of this residue in a biorefinery concept. However, its necessary further
489 studies in order to evaluate the actual potential of the utilisation of this residue since it
490 has been described that SE produce a higher abundance of β -O-4' linkages compared to
491 industrial lignin (Santos et al., 2015). Lignin-rich residue can be transformed into high
492 added value products such as building blocks for materials and fuels.

493 **3.5. Overall process material balance**

494 The diagram integrates the process of obtaining XOS (DP2-DP6) and bioethanol
495 from liquid and solid fractions, respectively, in figure 5. In the context of a biorefinery

496 process to valorisation, the barley straw residue is proposed to carry out a steam
497 explosion such as pretreatment, with the object to alter the structure. Concerning
498 pretreatment (180 °C 30 min), these conditions allowed an excellent recovery of
499 hemicellulose and cellulose in liquid fraction and solid fraction.

500 This paper has been a detailed process for obtaining XOS with a degree of
501 polymerisation DP2-DP6 from liquid fraction of barley straw pretreated through
502 enzymatic hydrolysis. 13.0 g of XOS per 100 g of raw material have been achieved
503 using the *Complex* cocktail enzymatic. After pretreatment, 54.1 g solid was recovered,
504 including 30.2 g of glucan, 5.6 g xylan and practically all lignin (16.7 g). High ethanol
505 yield from solid fraction was achieved using presaccharification and simultaneous
506 fermentation with high solid concentration (20% w/v). The final residue contained un-
507 hydrolysed cellulose and xylan fermentation and a high amount of lignin, which have
508 been characterised using FTIR technique (previous section).

509 The result indicated that BS could be used to obtain bioproducts such as
510 bioethanol, XOS, and a lignin-rich residue to revalue an agriculture residue.

511 *Fig. 5*

512 **4. Conclusions**

513 BS is a potential raw material to obtain bioethanol and value-added products
514 such as xylooligosaccharides for application in diverse industries. The strategy followed
515 in this study for obtained XOS was pretreated the raw material with SE following EH
516 through endo- β -(1,4)-D-xylanase and accessory enzymes were tested. The increment
517 production of XOS (DP2-DP6) with respect to liquid fraction of barley straw pretreated
518 was in the system *Maf*, *Complex*, commercial endo- β -(1,4)-D-xylanase, *NSAFA* was
519 56%, 64%, 55% and 53% respectively. Integration of bioethanol and bioproducts from

520 raw material could improve the economy of a possible biorefinery.

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524

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710 **Figure 2.** Profile of production of XOS during EH from *Lfbsp* with a) *Maf*
711 cocktail b) Complex cocktail

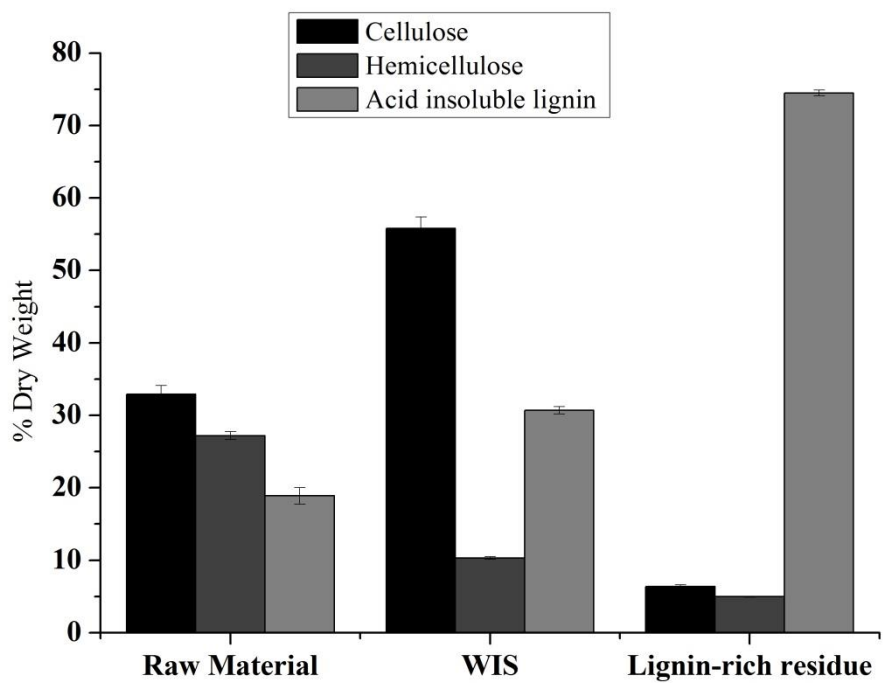
712 **Figure 3:** Profile of production of XOS during EH from *Lfbsp* with a) endo- β -
713 (1,4)-D-xylanase *NS50030* b) *NSAFA*

714 **Figure 4.** XOS (DP2-DP6) released in the different EH test

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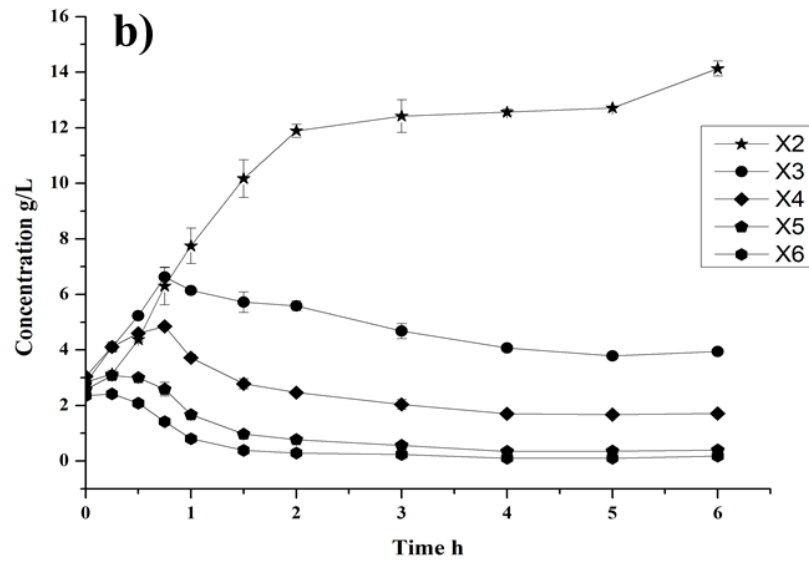
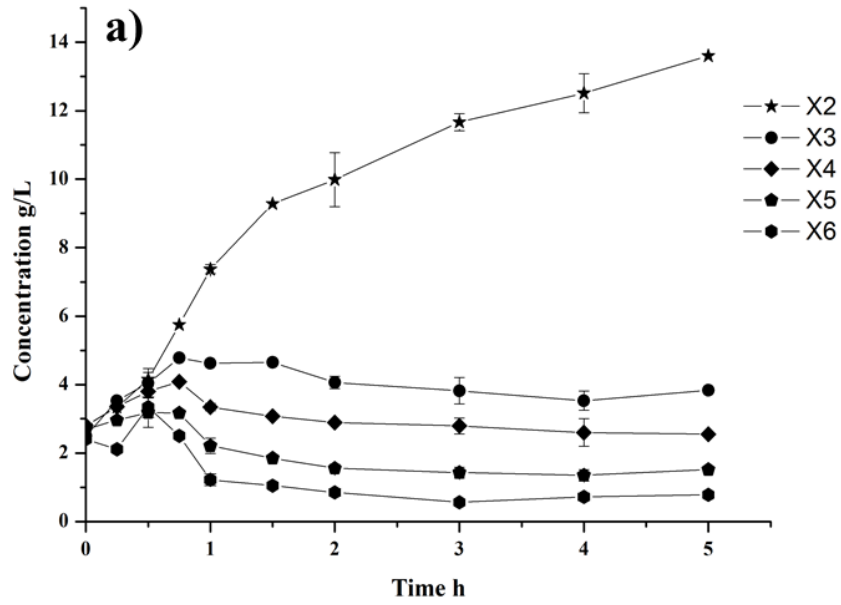
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720 **Figure 1.**

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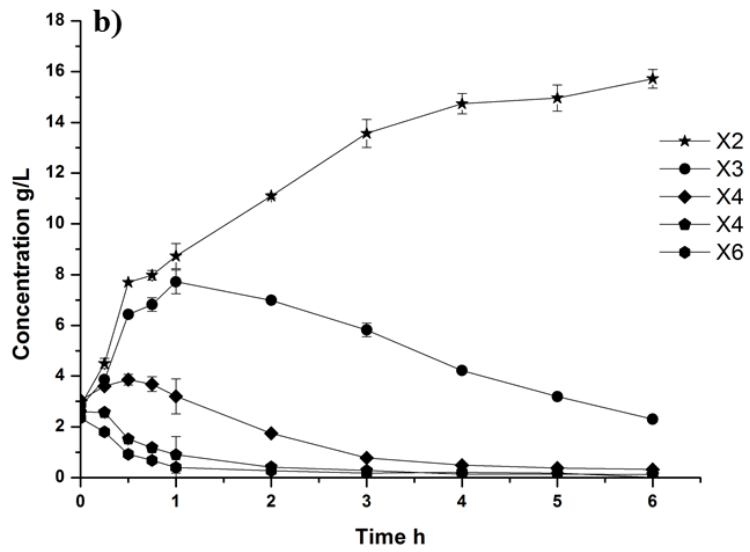
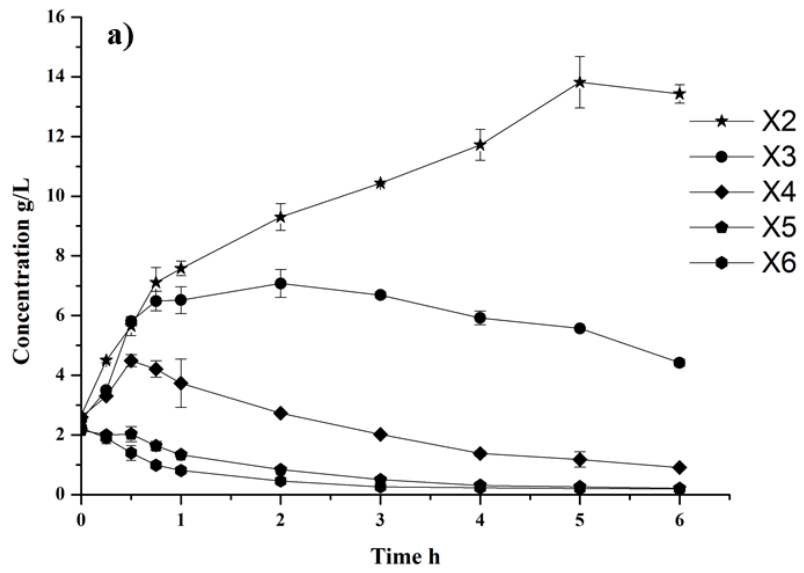
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Figure 2.

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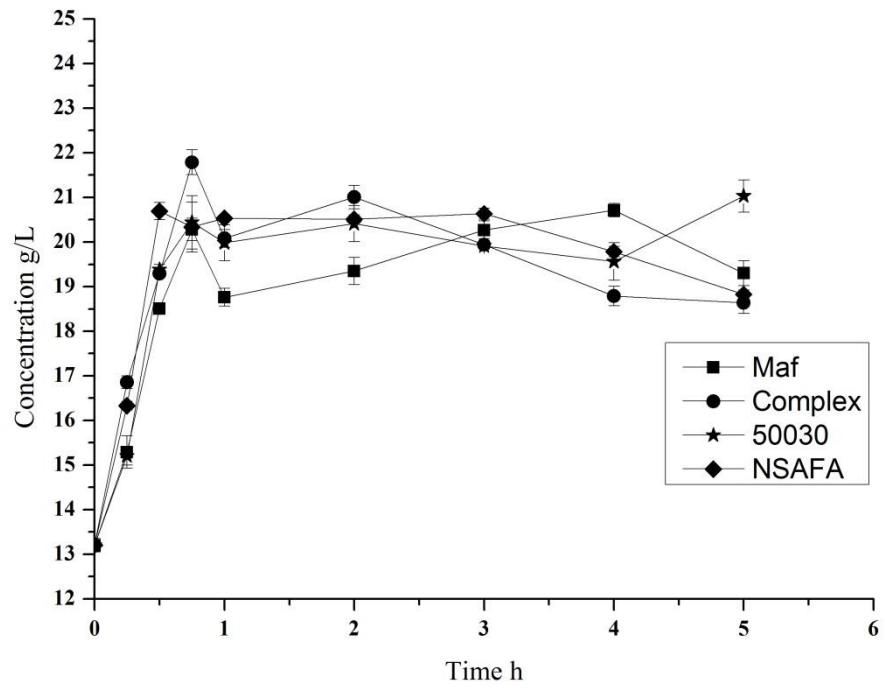
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Figure 3:



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729 **Figure 4.**

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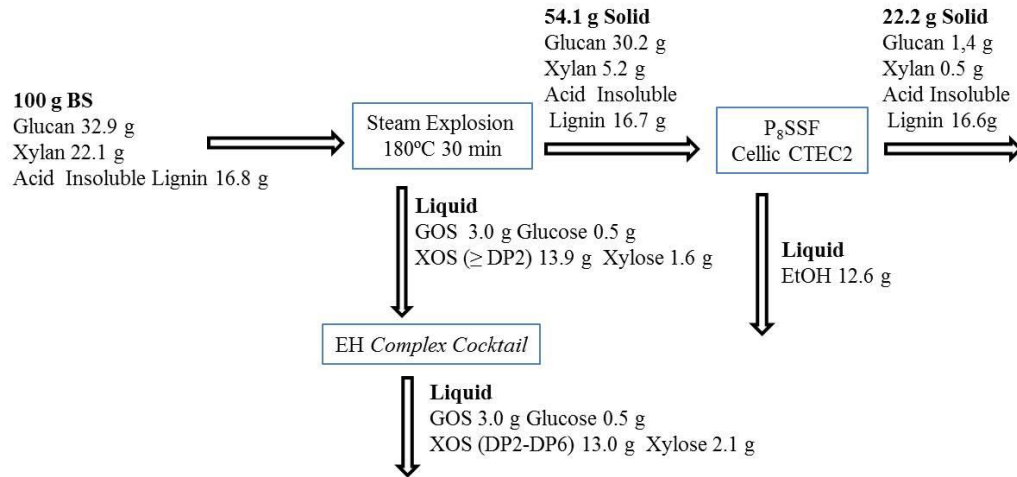
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737 Figure 5

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